

Option B — Quantum Physics and Nuclear Physics

- B1.** (a) (i) substitution into formula $E = \frac{hc}{\lambda}$
 to give $E = 4.48 \times 10^{-19} \text{ J}$; [1]
Units need not be stated.
- (ii) photon energy = 2.8 eV *or* work function = $7.36 \times 10^{-19} \text{ J}$;
 photon energy < work function;
hence no emission; [3]
- (b) electron emitted from surface will have energy $(2 \times 2.8 - 4.6) = 1.0 \text{ eV}$;
 photon can interact with an electron below surface;
 so energy is required to bring the electron to the surface;
 this energy is deducted from maximum kinetic energy of electron; [4]
- B2.** (a) γ -ray photons have discrete values of energy;
 which correspond to energy differences between energy states in nucleus; [2]
- (b) (i) ratio is $\frac{1}{2}$ so two parts Co, one part Ni *or* $2\text{Ni} = \text{Co}$;
 cobalt has two of three parts *i.e.* $\frac{2}{3}$ $2\text{Ni} + \text{Co} = N_0$ *i.e.* $\text{Co} = \frac{2}{3} N_0$; [2]
- (ii) $N = N_0 e^{-\lambda t}$;
 $0.67 = e^{-3\lambda}$ giving $\lambda = 0.133 \text{ yr}^{-1}$;
 use of $\lambda T_{\frac{1}{2}} = \ln 2$
 to give $T_{\frac{1}{2}} = 5.2 \text{ years}$; [3]
- B3.** (a) (electric) charge;
 strangeness;
 lepton number;
 parity;
 baryon number;
 angular momentum;
 isotopic spin; [3 max]
- (b) (i) lepton number / angular momentum is not conserved; [1]
- (ii) ${}^1_0n \rightarrow {}^1_1p + {}^0_{-1}e + \bar{\nu}$; [1]